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# Controlled microstructure and mechanical properties of Al<sub>2</sub>O<sub>3</sub>-based nanocarbon composites fabricated by electrostatic assembly method

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## Abstract

This work reports on the microstructure-controlled formation of interconnected carbon-layered Al<sub>2</sub>O<sub>3</sub> ceramics using carbon nanoparticles (CNP)-alumina (Al<sub>2</sub>O<sub>3</sub>) composite particles. The Al<sub>2</sub>O<sub>3</sub> micro-particles used in this study were obtained by granulation of nano-sized Al<sub>2</sub>O<sub>3</sub> nanoparticles with an average diameter of 150 nm. Then, CNP-Al<sub>2</sub>O<sub>3</sub> composite was fabricated using an electrostatic assembly method using the granulated Al<sub>2</sub>O<sub>3</sub> and CNP. The decoration of CNP on the surface of granulated Al<sub>2</sub>O<sub>3</sub> was investigated as a function of primary particle size and coverage percentage using a fixed amount of CNP. Notably, an interconnected layer of carbon particles at the interface of Al<sub>2</sub>O<sub>3</sub> that resemble the grain boundaries was obtained. The mechanical properties of the samples obtained with different particle size and CNP coverage on Al<sub>2</sub>O<sub>3</sub> particles were also investigated which presented the possibility to control the mechanical properties through microstructural design of composite ceramic materials.

**Keywords:** Nanocomposite, Electrostatic adsorption, Carbon microsphere, Alumina, mechanical property

## Introduction

It is well known that alumina (Al<sub>2</sub>O<sub>3</sub>) possesses good properties such as high hardness, excellent wear resistance, and high chemical stability. On the other hand, the drawbacks of alumina are its poor fracture toughness, low strength at elevated temperature as well as poor thermal shock resistance [1]. This has prompted intense research in alumina-based nanocomposite development at micro- and nano-scales. Functional ceramic composites with well-dispersed nano-size particles in the ceramic matrix are reported to improve not only mechanical properties such as failure strength, fracture toughness, fatigue, and wear resistance but also the electrical, magnetic, thermal, and optical properties [2–7]. In order to improve and control the mechanical properties of ceramics, microstructural porosity [8, 9], incorporation of additive fillers [10], and heat-treatment profiles [11, 12] have been used and reported. This shows that by controlling the microstructure of Al<sub>2</sub>O<sub>3</sub>, the desired

mechanical properties of Al<sub>2</sub>O<sub>3</sub> ceramics could be obtained. However, most reported work merely used the simple method of Al<sub>2</sub>O<sub>3</sub> powders mixing prior to sintering which is insufficient to obtain good control on the microstructure and design of Al<sub>2</sub>O<sub>3</sub> ceramics resulting in poor controllability of its mechanical properties. In the formation of nanocomposite by a conventional mixing method, it remains a huge challenge to obtain a homogenous decoration of nano-sized additive particles onto a designated primary particle due to the additive particles agglomeration. The uneven distribution caused by the agglomeration would then lead to adverse effects on the microstructural design as well as the properties of a ceramic composite. Therefore, a novel method via bottom-up assembly using an electrostatic adsorption method was used in this study to demonstrate the feasibility to obtain a good microstructural control and design that consequently allow controlled desired properties to be introduced into Al<sub>2</sub>O<sub>3</sub> ceramics such as optical, electrical, and mechanical properties. As one of the possible additives for Al<sub>2</sub>O<sub>3</sub> ceramic composites, various shapes of nano-sized carbon materials such as fiber (carbon nanotube (CNT), nanofiber) and plate-like (graphene) as well as

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particle have been developed. This enables the application of carbon-based materials as an additive for materials fabrication which has been reported recently.

In the development of carbon-based alumina composite, Kumari et al. reported the enhancement of thermal conductivity of carbon nanotube (CNT)-alumina composite up from 60 to 318% compared to pure alumina by changing the weight percentage of CNT addition and sintering temperature [4]. Besides that, owing to the exceptional tribological properties of carbon-based composite materials for applications such as power generation, transportation, and manufacturing, many researchers have focused their interest into the development of carbon-based composites [13, 14]. Ceramics with carbon reinforced surfaces have been reported to exhibit improved wear resistance and a reduced friction coefficient. Despite controversial reports on mechanical strength enhancement using carbon nanofiber (CNF) on alumina and zirconia, most authors have reported improvement in the mechanical properties. A recent study of CNT on the creep property of alumina drew an opposing conclusion as it is reported that depending on the addition amount of CNT, the creep strength could be either strengthened or weakened due to an impediment of grain boundary sliding or promotion of grain boundary diffusion or sliding, respectively [15]. Meanwhile, Crepo et al. reported that graphene oxide-reinforced alumina composite exhibits better creep resistance than CNF-reinforced alumina [16]. Also, due to the excellent lubricating properties of graphite, carbon-based materials are a good candidate for solid lubricant application. During dry friction, carbon-based composites are reported to generate a lubricating film from the exfoliation of carbon and its incorporation with the ceramic debris over the affected contact area [13]. However, most of the reported work involves the usage of sole mixing by either ultrasonic mixing of suspensions or a conventional mechanical milling, and no work has been demonstrated on the controlled decoration of carbon materials on ceramic leading to the formation of microstructure-controlled carbon-based ceramics. Therefore, in this study, CNP- $\text{Al}_2\text{O}_3$  composites were fabricated using electrostatic adsorption assembly which offers more controllability in its composite assembly and design. The  $\text{Al}_2\text{O}_3$  micro-particles used in this work were obtained using control granulation of nano-sized  $\text{Al}_2\text{O}_3$  particles. Then, the granulated  $\text{Al}_2\text{O}_3$  micro-particles obtained were used for the formation of carbon CNP- $\text{Al}_2\text{O}_3$  composite. The study was conducted systematically by varying the amount of carbon nanospheres from 0.3, 0.6, and 1.0 vol% (volume percent) and the average size of alumina particles used. The mechanical properties of carbon-based composite samples were then characterized and compared with a monolithic alumina sample using a three-point bending and indentation test. The inter-correlation between the microstructure

obtained and mechanical properties is also discussed and elucidated.

## Methods

Alumina nanoparticles with an average diameter of 150 nm were purchased from Taimei Kagaku Kogyo Co. and used as the precursor to obtain granulated  $\text{Al}_2\text{O}_3$  micro-particles. The granulation was carried out using spray drying of a mixture of  $\text{Al}_2\text{O}_3$  micro-particles with acrylic binder. The  $\text{Al}_2\text{O}_3$  micro-particles were then sieved to obtain three different average diameters 37, 62, and 98  $\mu\text{m}$  which were then used as the primary particles. CNP with an average diameter of 260 nm was purchased from Tokai Carbon Co. and used as the additive nanoparticles. As CNP is hydrophobic and not dispersible in a water medium, it was first dispersed in SDC (sodium deoxycholate) solution and then subjected to a hydrophilization treatment for the subsequent coating. Forty milliliters of a 0.1 wt% SDC solution was added to 1 g of CNP and dispersed by ultrasonication for 30 min. Then, the solution was centrifuged and washed three times using ion-exchange water which was carried out by stirring the water with a mixer. After that, modification of the surface charge was carried out using polycation and polyanion. Polydiallyldimethyl ammoniumchloride (PDDA) (average molecular weight 100,000 to 200,000, Sigma-Aldrich) and polysodium styrenesulfonate (PSS) as polyanion (average molecular weight 70,000, Sigma-Aldrich) were used as the polycation and polyanion, respectively. After that, the SDC-coated CNP were then alternately immersed into PDDA, PSS, and PDDA in order to induce a stable positive surface charge. After the adsorption process, the remaining suspension was dried and then collected. In the first investigation, the  $\text{Al}_2\text{O}_3$  particles with a diameter of 62  $\mu\text{m}$  were used and the feasibility of CNP coverage control on  $\text{Al}_2\text{O}_3$  particles was performed. The volume percentage of CNP added was 0.3, 0.6, and 1.0 vol%. In the investigation of the effect of  $\text{Al}_2\text{O}_3$  size, a fixed 0.6 vol% of CNP addition was set while  $\text{Al}_2\text{O}_3$  particles with different average diameters of 37, 62, and 98  $\mu\text{m}$  were used for the composite formation. The CNP- $\text{Al}_2\text{O}_3$  composite particles were first uniaxially pressed using a die with a diameter of 12 mm. The pressure applied was 300 MPa and the holding time was 5 min. After that, the pellet obtained was inserted into a graphite die with h-BN powder for hot-press sintering (Diavac Inc. Ltd.) The hot-press sintering was carried out in a vacuum atmosphere ( $8 \times 10^{-3}$  Pa) at 1350 °C (heating rate of 10 °C/min) for 2 h with a pressure of 30 MPa. The morphologies of the CNP- $\text{Al}_2\text{O}_3$  composites and the sintered microstructure obtained were observed using an S-4800 Field Emission Scanning Electron Microscope (FE-SEM, Hitachi S-4800). The zeta potential was measured using an Otsuka Electronics

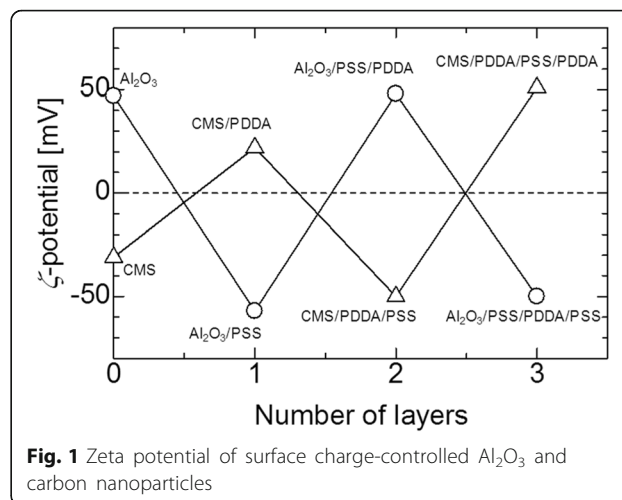
Co. Ltd., ELSZ-1 and Micro Tech Nission, ZEECOM Co. Ltd. As for the mechanical properties determination, the elastic modulus of the sample obtained was measured using a 3-point bending test. The sample was first cut into a strip-shaped test sheet and the dimension was fine-tuned using a surface grinder. The dimension of the test specimen prepared was  $3 \times 4 \times 40$  mm. After that, polishing was carried out using 0.5- $\mu\text{m}$  alumina and diamond paste with a grade of 30 and 9  $\mu\text{m}$ , respectively. The 3-point bending test was measured using an Instron type compact tester. First, the stress ( $\sigma$ ) was calculated using Eq. 1 where,  $l$ ,  $b$ , and  $h$  are the span distance and dimensions of each test piece, while  $P$  represents the load. Next, the relationship between stress and strain was plotted, and the elastic modulus was calculated from the slope of least squares. The crosshead speed was tested at 0.02 mm/min and the span at 30 mm.

$$\sigma = \frac{3P}{2bh} \quad (1)$$

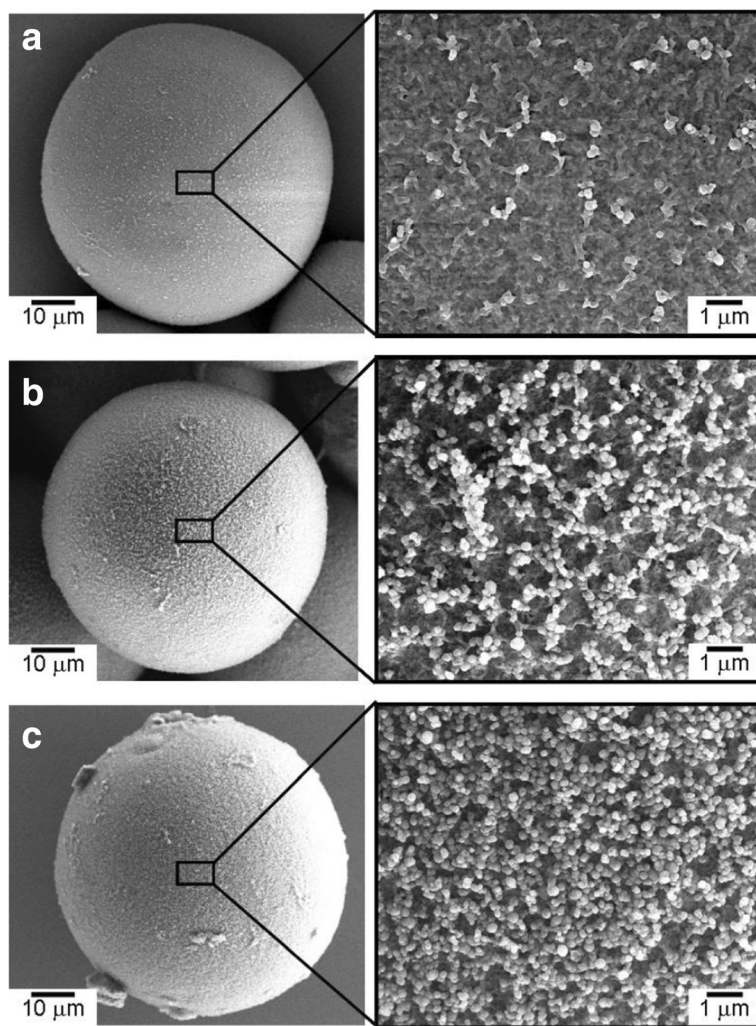
The hardness properties of the composite sample were further evaluated using indentation. The Rockwell indenter used consisted of a diamond ( $E_i = 1050$  GPa,  $\nu = 0.20$ ) with a nominal radius of curvature,  $R = 200$  embedded in a conical tip with an apex angle of  $120^\circ$ . The indenter was set in an Instron type tester (Sanwa Instruments) and was driven in at a crosshead speed of 0.05 mm/s to a fixed depth (20  $\mu\text{m}$ ). The load obtained during indentation was measured with a load cell (TCLZ-100KA, Tokyo Gakko), and the indent depth was measured with a non-contact electrostatic displacement meter (VE-222, Ono Sokki).

## Results and Discussion

Figure 1 shows the surface charge zeta potential of  $\text{Al}_2\text{O}_3$  particles and CNP after alternating coatings of PDDs and PSS, accordingly. It could be observed that the alumina and CNP exhibited a zeta potential of +55 and -55 mV, respectively, after three layers of coating. The achieved zeta potential after three layers of coating on both CNP and  $\text{Al}_2\text{O}_3$  micro-particles was stable. The surface morphologies of the CNP- $\text{Al}_2\text{O}_3$  composites with different 0.3, 0.6, and 1.0 vol% of CNP addition are shown in the SEM images of Fig. 2. From the higher magnification SEM images, it can be clearly observed that the amount of CNP that are adsorbed onto the surface of  $\text{Al}_2\text{O}_3$  particle increased with a higher volume percent of CNP addition. It is important to note the CNP are distributed homogeneously throughout the  $\text{Al}_2\text{O}_3$  surface without sign of agglomeration which portrays the advantage of the EA method to obtain an even and uniform distribution. The particle size of the CNP observed is approximately 260 nm. By fixing the addition of CNP at 0.6 vol% and varying the size of the  $\text{Al}_2\text{O}_3$  micro-particles from 37, 62, and 98  $\mu\text{m}$ , the distributions of CNP on the surface



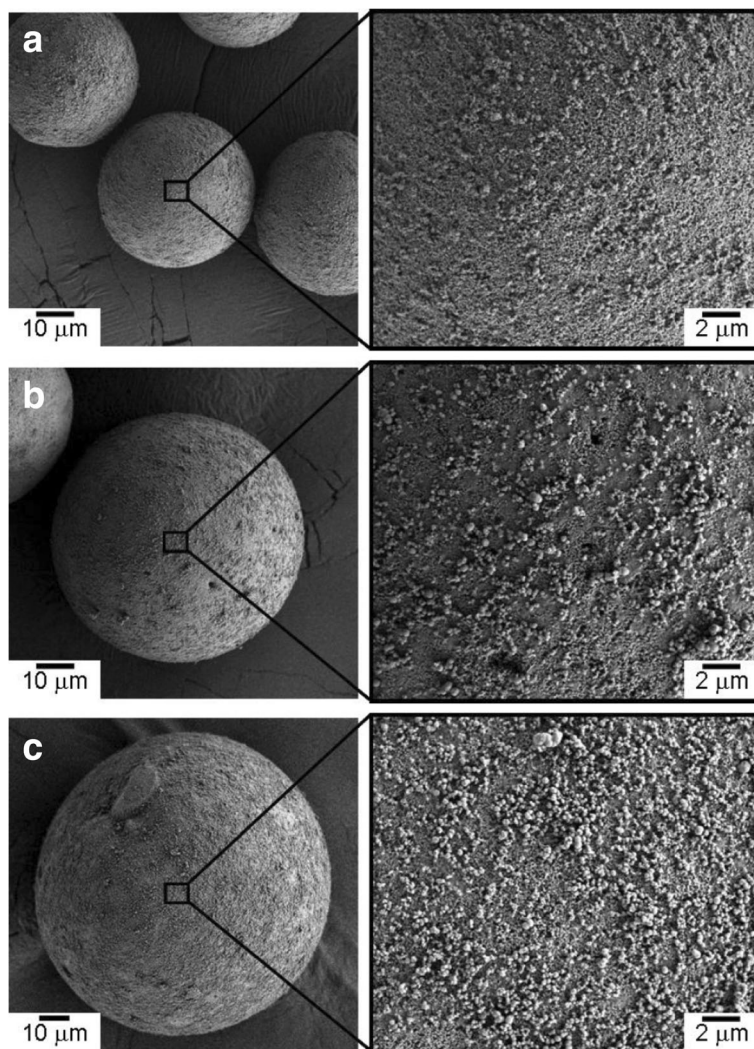
of  $\text{Al}_2\text{O}_3$  particle are shown in the SEM images in Fig. 3. From the observation of the SEM images, it could be seen that as the diameter of the particle size increased, the amount of CNP adsorbed on the surface was observed to increase accordingly. As larger  $\text{Al}_2\text{O}_3$  particles possess lower overall surface area compared to the smaller  $\text{Al}_2\text{O}_3$  particles, the collective surface area available for the adsorption of CNP is also lower compared to smaller  $\text{Al}_2\text{O}_3$  particles. Therefore, with a constant of 0.6 vol% addition of CNP, a higher amount of CNP was adsorbed onto the lower overall surface area of larger  $\text{Al}_2\text{O}_3$  particles. As a result, the amount of CNP adsorbed onto the  $\text{Al}_2\text{O}_3$  surface is observed to increase with the diameter size of  $\text{Al}_2\text{O}_3$  particles that led to a higher density of CNP adsorption onto the surface of 98  $\mu\text{m}$   $\text{Al}_2\text{O}_3$  particles. On the other hand, as the particle size reduced, the available overall surface area accessible for CNP adsorption on  $\text{Al}_2\text{O}_3$  increased and therefore, a sparse distribution of CNP is observed due to the insufficient amount of CNP in the suspension (at a fixed 0.6 vol%). The sintered microstructure obtained using the CNP- $\text{Al}_2\text{O}_3$  composite and high magnification at the interface as shown in Fig. 4. From the SEM image in Fig. 4a, it can be seen that the microstructure obtained reflects the shape of the obtained CNP- $\text{Al}_2\text{O}_3$  composite. It is noteworthy that the grain boundaries are connected forming a network along the grain boundaries. From the observation of the CNP network that is formed along the grain boundaries, the homogeneity of CNP distribution on the surface of  $\text{Al}_2\text{O}_3$  particles can be determined. This result shows that it is feasible to obtain a microstructure-controlled composite material by designing the composite precursor. From the higher magnification SEM image in Fig. 4b, the presence of a carbon layer in between the interface of the  $\text{Al}_2\text{O}_3$  grain boundary can be observed. This shows that the sintering of the CNP in between the  $\text{Al}_2\text{O}_3$  particles during hot-press sintering led to the formation of an even coating of a carbon layer along the grain boundaries. It is also important to note



**Fig. 2** SEM images of the **a** 0.3 vol%, **b** 0.6 vol%, and **c** 1.0 vol% CNP coated on  $\text{Al}_2\text{O}_3$  granulation particle with average diameter of 62  $\mu\text{m}$

that the  $\text{Al}_2\text{O}_3$  matrix obtained is dense and well sintered with no observation of pores as shown in Fig. 4b. This is due to the formation of densely packed granulated  $\text{Al}_2\text{O}_3$  nanoparticles (150 nm) which allows good sinter-ability that demonstrated the novel technique of this work. The elastic moduli of the CNP- $\text{Al}_2\text{O}_3$  composite obtained using a 3-point bending test plotted as a function of  $\text{Al}_2\text{O}_3$  particle size and surface coverage percentage are shown in Fig. 5. From Fig. 5a, the exhibited elastic modulus of the sample fabricated using  $\text{Al}_2\text{O}_3$  particles only is approximately 390 GPa which is consistent with the results reported on polycrystalline  $\text{Al}_2\text{O}_3$  which is between 300 and 400 GPa [6, 12]. The achievement of this elastic modulus value corroborates with the SEM observation where a good microstructure and compaction was achieved using granulated  $\text{Al}_2\text{O}_3$  nanoparticles. In the study of Ashizuka et al. on the effect of porosity on the mechanical properties of alumina ceramics, the elastic modulus of the ceramic without porosity (0%) is slightly lower at approximately 380 GPa [17]. As for the elastic

moduli of the CNP- $\text{Al}_2\text{O}_3$  composites, it can be seen that the property could be controlled as it decreased linearly with either higher volume percent addition of CNP or increment in the  $\text{Al}_2\text{O}_3$  particle size. A similar trend was also observed in the work of Shin et al., where the elastic moduli of their reduced graphene oxide and single-wall CNT-alumina composites were reduced by increasing the additive content [6]. As both factors (amount of CNP and particle size of  $\text{Al}_2\text{O}_3$ ) highly influence the specific surface area and led to greater adsorption of CNP on the  $\text{Al}_2\text{O}_3$  particle surface, this would inhibit the sintering of  $\text{Al}_2\text{O}_3$  and a possible slipping effect of the carbon layer resulted in lower elastic modules [6]. This finding is consistent with those reported by Gopalan et al. where the CNT used in their composite retarded grain growth but had no effect on the grain boundary sliding resulting in the occurrence of superplasticity [15]. This finding indicates the possibility to alter and control the elastic modulus of an  $\text{Al}_2\text{O}_3$  ceramic by controlling the microstructural formation via the design of the precursor composite

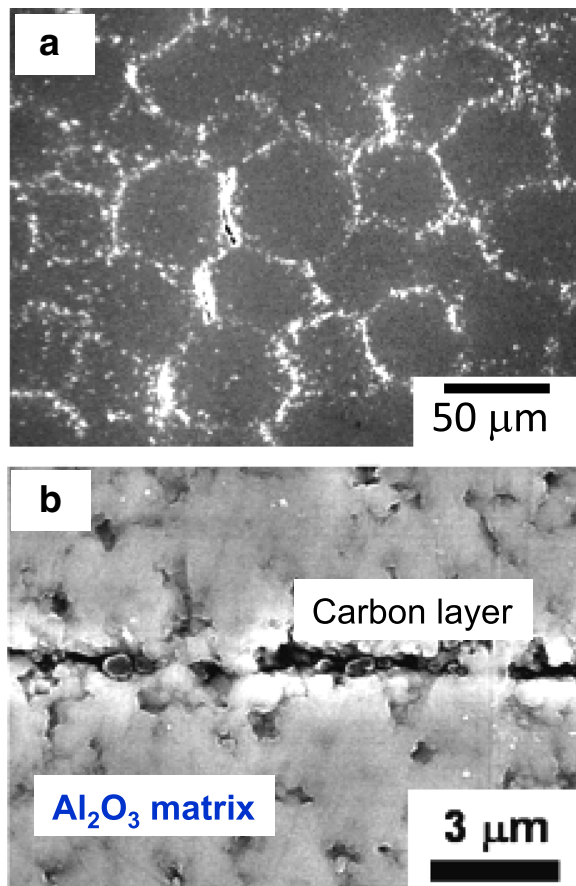


**Fig. 3** SEM images of the 0.6 vol% CNP coated on  $\text{Al}_2\text{O}_3$  granulation particles with average diameter of **a** 37, **b** 62, and **c** 98  $\mu\text{m}$

used in the formation of  $\text{CNP-Al}_2\text{O}_3$ . In Fig. 5b, the plot of the obtained elastic moduli as a function of the CNP coverage ratio on  $\text{Al}_2\text{O}_3$  is shown. A linear correlation between the CNP coverage ratio and elastic modulus strength is observed which further corroborate with the abovementioned results. Therefore, from these results, it is demonstrated that the mechanical properties of a  $\text{CNP-Al}_2\text{O}_3$  composite ceramic can be controlled via the CNP coverage ratio by either altering the amount of CNP addition or the particle size of primary  $\text{Al}_2\text{O}_3$ . In the determination of the micro-hardness of the  $\text{CNP-Al}_2\text{O}_3$  composite samples, a comparison between pure  $\text{Al}_2\text{O}_3$  and  $\text{CNP-Al}_2\text{O}_3$  samples fabricated with 1.0 vol% of CNP addition with different  $\text{Al}_2\text{O}_3$  particle sizes of 37, 62, and 98  $\mu\text{m}$  was undertaken. The indentation results obtained are shown in Fig. 6. The results obtained show that the pure alumina sample exhibited the highest hardness value while the hardness of  $\text{CNP-Al}_2\text{O}_3$  composite samples reduced with larger  $\text{Al}_2\text{O}_3$  particle size. This is due

to the lower overall surface area of  $\text{Al}_2\text{O}_3$  when the particle size increase leading to a higher amount of CNP adsorbed on the surface. Subsequently, the higher amount of CNP on the  $\text{Al}_2\text{O}_3$  interface led to reduced hardness due to either the inhibition of an effective sintering between the  $\text{Al}_2\text{O}_3$  interface or the slip of the continuous connected carbon layer along the grain boundaries of  $\text{Al}_2\text{O}_3$ . Therefore, it is crucial to have a controlled distribution of CNP on the surface of  $\text{Al}_2\text{O}_3$  in order to induce the formation of a desired microstructure leading to the desired mechanical properties of  $\text{CNP-Al}_2\text{O}_3$  composite.

From the indentation load ( $P$ )-depth ( $h$ ) curve ( $P$ - $h$  curve) during the loading and unloading cycle, microstructural change and mechanism of surface deformation could be obtained [18]. The simple quadratic equation involving indentation load ( $P$ ) and penetration depth ( $h$ ) shown in Eq. 2 can be used for the analysis of loading process [18–20].



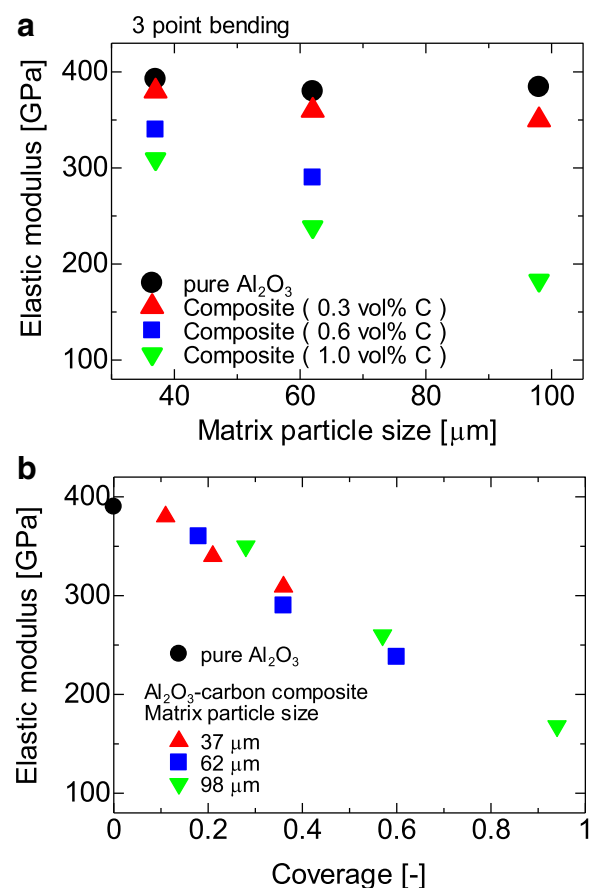
**Fig. 4** **a** Microstructure of 0.6 vol% CNP- $\text{Al}_2\text{O}_3$  composite using  $\text{Al}_2\text{O}_3$  with the average diameter of 62  $\mu\text{m}$ . **b** Grain boundary of CNP- $\text{Al}_2\text{O}_3$  composite. Carbon layer could be observed at the interface between the  $\text{Al}_2\text{O}_3$  matrix

$$P \propto h^2 \quad (2)$$

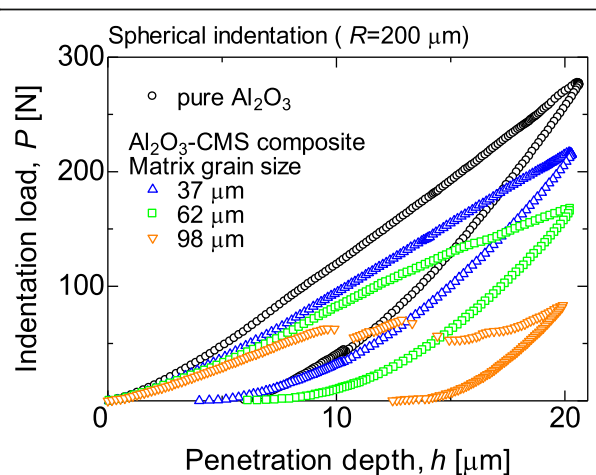
The loading of monolithic  $\text{Al}_2\text{O}_3$  demonstrated a correlation with the  $P$ - $h$  curve similar to the quadratic Eq. 2 while CNP- $\text{Al}_2\text{O}_3$  composite fabricated using  $\text{Al}_2\text{O}_3$  particle with the size of 37 and 62  $\mu\text{m}$  demonstrated a linear with deviated curve from the monolithic  $\text{Al}_2\text{O}_3$  loading curve, respectively. This indicates the presence of CNP within the microstructure (at the grain boundary interface) which resulted in local deformation along the grain boundaries. As for CNP- $\text{Al}_2\text{O}_3$  composite fabricated using  $\text{Al}_2\text{O}_3$  with the particle size 98  $\mu\text{m}$ , the high density of CNP at the grain boundaries resulted in discontinuity of  $P$ - $h$  hysteresis curve and demonstrated the lowest hardness due to occurrence of grain boundary slip or surface microfracture.

## Conclusions

In this work, a feasible controlled formation of CNP- $\text{Al}_2\text{O}_3$  composite by an electrostatic adsorption method is demonstrated. The  $\text{Al}_2\text{O}_3$  micro-particles used were obtained by



**Fig. 5** Elastic moduli of CNP- $\text{Al}_2\text{O}_3$  composites as a function of **a** matrix particle size and **b** CNP coverage on  $\text{Al}_2\text{O}_3$  particles



**Fig. 6** Hysteresis curves of indentation load and penetration depth of 1.0 vol% CNP- $\text{Al}_2\text{O}_3$  composites

granulation of nano-sized (150 nm)  $\text{Al}_2\text{O}_3$  particles which enabled better compaction and sinter-ability. In the formation of composite ceramics, parameters involving the amount of CNP (0.3, 0.6, 1.0 vol%) and primary granulated  $\text{Al}_2\text{O}_3$  micro-particle sizes (37, 62, 92  $\mu\text{m}$ ) were investigated. It is demonstrated that by controlling the amount of CNP additives and  $\text{Al}_2\text{O}_3$  micro-particle size, different surface coverage could be obtained leading to controlled microstructure formation with different mechanical properties. Using the homogenous CNP- $\text{Al}_2\text{O}_3$  composite, a continuous interconnected carbon layer was obtained along the grain boundaries of  $\text{Al}_2\text{O}_3$ . A dense and compact  $\text{Al}_2\text{O}_3$  matrix was also observed due to the good sintering of  $\text{Al}_2\text{O}_3$  nanoparticles. From the results of a 3-point bending and indentation test, the control of mechanical properties was demonstrated by adjusting the coverage of CNP on  $\text{Al}_2\text{O}_3$ . The change in elastic modulus was either due to the inhibition of effective sintering or the slipping of the carbon layer generated at the  $\text{Al}_2\text{O}_3$  interface. From this study, we have demonstrated the feasibility of ceramics microstructural design with an interconnected interface using CNP- $\text{Al}_2\text{O}_3$  composite. This method of microstructural design will open up greater possibilities and potential for materials design through bottom-up assembly to induce the desired properties for a wide range of applications.

#### Abbreviations

CNP: Carbon nanoparticles; PDPA: Polydiallyldimethyl ammoniumchloride; PSS: Polysodium styrenesulfonate; SDC: Sodium deoxycholate; SEM: Scanning electron microscope

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#### Authors' Contributions

HM and WKT designed the study and contributed to the manuscript writing. NH, AY, GK, and AM provided technical and scientific insight and contributed to the editing of the manuscript. All authors read and approved the final manuscript.

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#### Availability of Data and Materials

All data generated or analyzed during this study are included in this published article (and its supplementary information files).

#### Competing Interests

The authors declare that they have no competing interests.

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